AN EFFICIENT SYNTHESIS OF THE NAPHTHALENE MOIETY OF NEOCARZINOSTATIN CHROMOPHORE

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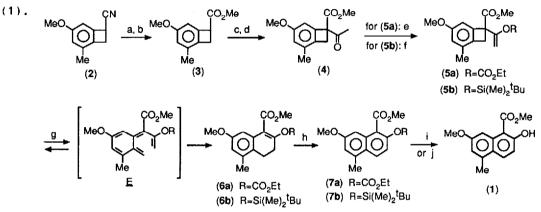
Abstract: Methyl 2-hydroxy-7-methoxy-5-methyl-1-naphthalenecarboxylate, the methanolysis product of neocarzinostatin chromophore, was efficiently synthesized from 1-cyano-5-methoxy-3-methylbenzocyclobutene employing an electrocyclic reaction of <u>E-o</u>-guinodimethane as the key step.

Neocarzinostatin chromophore (NCS-chr.), non-protein component of the antitumor antibiotic neocarzinostatin isolated from the culture filtrate of <u>Streptomyces carzinostaticus</u> var. F-41 by Ishida<sup>1</sup>, has attracted considerable interest among synthetic organic chemists due to its intriguing structural features.<sup>2</sup> Our efforts in this area have focussed on construction of the naphthalene moiety (1)<sup>3</sup>, which was obtained by methanolysis of NCS-chr., possessing four different kinds of functionalities on the naphthalene ring. The chemical objectives of this total synthesis endeavor were not only to establish a novel and practical synthetic route for (1) but also to examine whether the highly selective electrocyclic reaction <u>via</u> <u>E-o</u>-quinodimethane, generated <u>in situ</u> by the thermolysis of 1-alkeny1-1-carbomethoxybenzocyclobutene<sup>4</sup>, for constructing the naphthalene ring system would succeed with fully functionalized benzocyclobutenes, i.e. (5)  $\rightarrow$  (6).

The syntheses of (5a,b), the key intermediates in the experimental verification of this approach, start with 1-cyano-5-methoxy-3-methylbenzocyclobutene  $(2)^5$ , which is readily prepared from 4-methoxy-2-methylbenzaldehyde<sup>6</sup> by a standard five step sequence in 62 % overall yield. Hydrolysis of (2) and esterification of the resulting carboxylic acid provided the methyl ester (3) in 93 % yield. Treatment of (3) with lithium diisopropylamide (LDA) and acetaldehyde followed by oxidation with the conditions of Swern provided (4) in 86 % yield. Conversion of (4) into the key intermediate (5a) was achieved by treatment with LDA and ethyl chloroformate in 80 % yield. Heating of the enol carbonate (5a) at 180°C for 5 min in <u>o</u>-dichlorobenzene yielded the expected dihydronaphthalene (6a), as a sole product, in 79 % yield. Subsequent treatment with 2,3-dichloro-5,6-dicyano-1,4-benzoquinone (DDQ) in refluxing toluene caused a rapid and clean dehydrogenation to provide the

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naphthalene (7a) in quantitative yield. Finally, (7a) was converted by alkaline hydrolysis into the desired crystalline (1), mp 109-110°C (lit.<sup>3</sup>, mp 104-105°C), in 97 % yield. Alternatively, treatment of (4) with <sup>t</sup>butyldimethylsilyl trifluoromethanesulfonate and triethylamine<sup>7</sup> provided the silyl enol ether (5b), which was immediately heated at 180°C in <u>o</u>-dichlorobenzene for 1 h followed by oxidation of the resulting (6b) with DDQ in a one-pot operation to give the corresponding naphthalene (7b). Desilylation of the crude (7b) with tetrabutylammonium fluoride proceeded smoothly to afford (1) in 46 % overall yield from (4). The material thus obtained was identical both chromatographically and spectroscopically (<sup>1</sup>H nmr, ir, mass) with an authentic sample of



Reagents a, KOH, aq.EtOH; b, AcCl, MeOH; C, LDA, CH<sub>3</sub>CHO; d, (COCl)<sub>2</sub>, DMSO, NEt<sub>3</sub>;
e, LDA, CICO<sub>2</sub>Et; f, <sup>t</sup>Bu(Me)<sub>2</sub>SiOTf, NEt<sub>3</sub>; g, 180°C, *o*-dichlorobenzene; h, DDQ;
i, K<sub>2</sub>CO<sub>3</sub>, aq.MeOH; j, Bu<sub>4</sub>NF

The synthesis reported here is a practical method for the preparation of naphthalene moiety of NCS-chr. since it requires eight steps from the literature known benzocyclobutene (2) and proceeds in 51 % (via enol carbonate) overall yield. Of equal importance, this synthesis demonstrates that the electrocyclic process via  $\underline{E}$ -o-quinodimethane is also predominant during the thermolyses of highly functionalized benzocyclobutenes.

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## References and Notes

1) N. Ishida, K. Miyazaki, K. Kumagai, and M. Rikimaru, <u>J. Antibiotics, Ser.</u> <u>A-18</u>, 68 (1965). 2)K. Edo, M. Mizugaki, Y. Koide, H. Seto, K. Furihata, N. Otake, and N. Ishida, <u>Tetrahedron Lett.</u>, **26**, 331 (1985). 3)K. Edo, S. Katamine, F. Kitame, N. Ishida, Y. Koide, G. Kusano, and S. Nozoe, <u>J. Antibiotics</u>, **33**, 347 (1980); Structure revision and the total synthesis, see M. Shibuya, K. Toyooka, and S. Kubota, <u>Tetrahedron Lett.</u>, **25**, 1171 (1984). 4)K. Shishido, H. Komatsu, K. Fukumoto, and T. Kametani, <u>Chem. Lett.</u>, 2117 (1987). 5)T. Kametani, Y. Kato, T. Honda, and K. Fukumoto, <u>J. Chem. Soc.</u>, <u>Perkin</u> <u>Trans. 1</u>, 2001 (1975). 6)4-Methoxy-2-methylbenzaldehyde was easily synthesized from methyl 4-hydroxy-2-methylbenzoate, prepared by the method of Rubottom [G. M. Rubottom and D. S. Krueger, <u>Tetrahedron Lett.</u>, 611 (1977).], <u>via</u> sequential methylation (MeI, K<sub>2</sub>CO<sub>3</sub>, acetone), reduction (LiAlH<sub>4</sub>), and oxidation (PCC) in 88 & overall yield. 7)L. N. Mander and S. P. Sethi, <u>Tetrahedron Lett.</u>, **25**, 3 (1984).